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Effects of Urbanization on environmental parameters in aquatic systems along an urban-rural gradient in Northeastern Illinois

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Abstract

The transformation of landscapes from rural to urban land use impacts affected ecosystems. Replacing natural vegetation with impervious surfaces and introducing environmental stressors through increased human activity can significantly alter aquatic ecosystems. This study examined the effects that urbanization had on salinity, dissolved oxygen, temperature, conductivity, and pH of aquatic systems along an urban-rural gradient in Illinois, Chicago land area- Little Calumet River, Thorn Creek, and Kankakee River. The first objective of this project was to examine whether there were changes or fluctuations among various environmental parameters that may be due to heavily populated residential areas when compared to a preserve. The second objective compared the responses of the environmental parameters against the urban- rural gradient. The study was conducted from February to August 2010. Data collection was from March to May 2010. There was a significant difference in all variables along the urban-rural gradient based on location (p<0.001), date (p=0.001), and location date interaction (p=0.001). Preserve sites had significantly higher salinity values than residential sites (p=0.029). The values for examined parameters in the study ranged 3.45-17.10°C (temperature), 73.73-112.03 % saturation (dissolved oxygen), 352.33-1113.33 µS (conductivity), 0.20-0.50ppt (salinity), and 7.80-8.90 (pH) in residential areas. While values ranged 3.60-17.23°C (temperature), 42.67-104.23 % saturation (dissolved oxygen), 352.95-1299.57 µS (conductivity), 0.10-0.80ppt (salinity), and 7.20-8.87 (pH) in preserve areas respectively. On average, along the urban-rural gradient there was a decrease in conductivity and salinity in residential areas. Average values for parameters in the Little Calumet River residential site were 10.34±5.55 °C (temperature), 8.18±0.28 (pH), 900.67±162.38 µS (conductivity), 0.49±0.03 ppt (salinity), 86.46±9.29 % saturation (dissolved oxygen). Average values for parameters in Thorn Creek residential site ranged 8.93±4.10°C (temperature), 8.49±0.32 (pH), 522.11±154.60µS.
(conductivity), 0.34±0.05ppt (salinity), 94.50±12.31 % saturation (dissolved oxygen). Average values for parameters in the Kankakee River residential site ranged 11.23±3.49 °C (temperature), 8.27±0.32 (pH), 531.03±102.37 µS (conductivity), 0.29±0.04ppt (salinity), 86.99±7.36 % saturation (dissolved oxygen). On average, along the urban-rural gradient there was an increase in dissolved oxygen, and a decrease in conductivity and salinity. Average values for parameters in the Little Calumet River preserve sites ranged 11.36±3.64 °C (temperature), 8.08±0.51 (pH), 952.57±186.14 µS (conductivity), 0.59±0.13 (salinity), 71.70±21.21 % saturation (dissolved oxygen). Average values for parameters in Thorn Creek preserve sites ranged 8.93±3.99°C (temperature), 8.37±0.26 (pH), 612.68±212.83 µS (conductivity), 0.39±0.09ppt (salinity), 86.91±4.45 % saturation (dissolved oxygen). Average values for parameters in the Kankakee River preserve site ranged 11.27±3.08 °C (temperature), 8.49±0.24 (pH), 473.90±96.10µS (conductivity), 0.27±0.07ppt (salinity), 87.10±4.74% saturation (dissolved oxygen). The analysis emphasize that aquatic systems are the product of their land, and the influence of aquatic systems is driven by human pressures. The findings indicate that efforts should be stressed to preserve and improve land in order to achieve a hydrological balance, thus creating a more habitable aquatic ecosystem for aquatic plants and animals, and humans.

**Keywords:** Urban-Rural Gradient; Salinity; Dissolved Oxygen; Temperature; Conductivity; Aquatic Ecosystem; population density; sustainable land use; pollution; environmental parameters
1. Introduction

The disturbing impacts of urbanization on natural landscapes and habitats are expected to continue to increase rapidly into the future (United Nations Human Settlements Programme, 2004). The process of urbanization can generate artificial and impervious surfaces (McDonnell and Hahs, 2008). Urbanization, which is driven by human density, ultimately results in new land cover types and biotic assemblages of plants and animals. In turn, this can alter the types and frequency of disturbance regimes (McDonnell and Pickett, 1993; Kinzig and Grove, 2001). Urbanization has the ability to alter the flux of nutrients, organisms, energy, and water within and between landscapes (McDonnell and Hahs 2008).

Urbanization can cause habitat loss, changes in hydrology, increases in sedimentation and pollution of water bodies, and changes in soil properties (Hamer and McDonnell 2008). Replacing natural vegetation with impervious surfaces and introducing environmental stressors through increased human activity can significantly alter aquatic ecosystems. Urban aquatic systems suffer the most damage due to high population densities and pollution. Over 476,000 ha of arable land are lost annually by the expansion of urban areas (Pouyat et al., 2002). Urban areas such as Chicago have approximately 386 people/km², suburban locals such as Park Forest have approximately 193 people/km², while rural areas such as Bourbonnais have population densities less than 193 people/km² (United States Census Bureau 2010). In most cases urban land conversion results in poor conditions for plant growth, with soil suffering from altered nutrient
fluxes or lacking essential nutrients (Pouyat et al., 2001). Previous studies have shown that soil has a large capacity for retaining hydrophobic chemicals, which can alter soil nutrient levels (Wong et al., 2004). Recent studies have found high conductivity (i.e., the presence of dissolved metals and salts in water) to negatively affect animal species populations in urban areas (Hamer and McDonnell 2008). Studies have shown that animal species like amphibians had lower survival, growth, and development rates due to nitrogen pollution and heavy metals in urban areas (Boone and Bridges, 2003; Casey et al., 2005; Massal et al., 2007). Blair (1996) reported a decrease in species richness with urbanization in Coastal California, in addition to species being methodically replaced along a more urbanized gradient.

Development of cities and small towns creates environmental conditions that can be studied using gradient analysis (McDonnell and Hahs 2008). More than 50% of the world’s population resides in urban areas (World Resources Institute, 1996). Studies have shown waste incineration, vehicle exhaust, and cigarette smoke to be the main sources of urban pollution. Urban areas are in close proximity with these sources; and urban residents have higher health risks. Knowledge of our residential environmental parameters should be a major concern because where one lives affects his or her educational opportunity, quality of life, and access to health care (Billard 1993). Research on urbanization allows scientists the opportunity to survey how natural communities are structured and function under dramatically altered landscapes (Blair and Johnson 2008). Also, environmentally sustainable land use and conservation are necessary to avoid species extinction (Blair and Johnson 2008). It is apparent that urbanization disrupts our
natural ecosystems in addition to posing a major threat to biota. More ecological information would help increase the number of inhabitable cities and help achieve better research, management, restoration, and conservation outcomes (McDonnell and Hahs 2008).

Two related variables are water temperature and dissolved oxygen. Water temperature and dissolved oxygen are inversely related environmental variables crucial to the survival of aquatic organisms and their ability to resist certain pollutants. Large industrial businesses such as power plants use water for cooling purposes and released water can affect downstream habitats. Low levels of dissolved oxygen indicate eutrophication in which water quality is reduced and organisms begin to die. Also associated with eutrophication is high turbidity, during which sunlight cannot pass through dense and gray colored water, causing excess organic material and low levels of oxygen. Phosphorous is a pollutant that causes eutrophication of aquatic systems (Bennett 2003). Soil phosphorous moves downhill into aquatic systems from sorbed to eroded soil particles and are dissolved in surface runoff and groundwater (Bennett 2003).

Also related to dissolved oxygen are water pH, conductivity and salinity. The pH of the water can be affected by chemicals in the water. It is important that the pH remains stable and within certain limits for organisms to survive and thrive. Jennifer et. al (2006) reported that changes in pH values resulted in changes in species abundance and interactions with other organisms. The amount of dissolved solids in the water impacts its ability to conduct an electrical current. High specific conductance can affect the suitability of water for domestic, industrial, and
agricultural use. High levels of conductance may affect the taste and odor of drinking water and may cause gastrointestinal distress. Likewise, high salinity may inhibit survival of nearby crops. Impervious surfaces may contribute to nutrient loss, high salinity, high conductivity, and changes in pH and dissolved oxygen (Environmental Protection Agency 2007).

Similarly, another study reported that anthropogenic changes in water salinity, conductivity, dissolved oxygen, and temperature affected species survival and interspecific interactions (U.S. Geological Survey 2005). Comparison of aquatic systems in urban areas with surrounding rural areas should help identify trends in environmental variables associated with decline in water quality. In addition, regional aquatic systems can provide a baseline against which potentially more intense human activity in urban areas can be judged (Cannon and Horton 2009).

Previous studies that examined larger areas are too broad to capture the specific historical patterns which have developed within these areas. Many studies provide information of how animals are affected, but have no underlying contextual dynamics that explain how their observations came to be. The precise magnitude of urbanization can only be measured by the environments response, which is why air, soil, and water quality testing are vital. There is an increasing awareness of problems caused by environmental degradation, due to increasing demands placed on the environment by population growth. A study of some key physical and chemical characteristics from the Little Calumet River, Thorn Creek, and the Kankakee River will help expand local knowledge of these aquatic systems, and is also essential for effective
management and restoration policies and procedures. This research addresses water quality traits related to surrounding land cover of natural and developed aquatic ecosystems. In addition, the study examines how environmental parameters vary across different land-use types along the urban-rural gradient.

The overall object of this research investigated the effects that urbanization had on environmental variables including salinity, dissolved oxygen, temperature, conductivity, and pH, along an urban-rural gradient from Chicago to rural Kankakee (Fig. 1). Changes in environmental parameters may impact plant and animal life. This research provides information of a gradient that can be used to establish basic ecological research in areas where previous information is absent. This research also provides a common element that allows for a greater integration between studies conducted at different places. In this study, the southern part of Chicago is designated as the high density urban area, Park Forest is south of Chicago and is the less heavily populated suburban area, and Bourbonnais located south of Park Forest is the least heavily populated and is designated the rural area.

Fig 1. of Urban-Rural Gradient Map
This study examined how urbanization affects water quality. I hypothesized that urbanization would have an effect on the parameters and that changes will be noted along the urban rural gradient. It is expected that each city will have its own characteristic aquatic signature determined by the history of land use. It was expected that the urban end of the gradient, Little Calumet River and residential plots will fail to meet EPA criteria for water quality standards. The changes are indicative of harmful increases in salinity, and decreases in dissolved oxygen which may disrupt aquatic plant and animal processes. Higher levels of salinity and conductivity and lower levels of dissolved oxygen were found at the Little Calumet River and did not meet EPA criteria. The Little Calumet River is located within the urban area of Chicago. Water bodies near heavily populated areas may be less suitable for species survival than water bodies near preserves. The Chicago city area provides an excellent opportunity for studying the effects of urban-rural land-use gradient on water quality due to its vast economic and industrial history.

2. Materials and Method

2.1. Study Area

The study was conducted in the Chicago City and surrounding suburban and rural area. The urban-rural gradient included water bodies near three cities in Illinois. All plots can be accessed from Interstate 57 which runs North and South. The city of Chicago and its surrounding suburban and rural areas have an industrial, environmental and demographic history. Chicago is located in Cook County, Illinois and houses approximately 9.7 million human residents (United
States Census Bureau 2010), and is the third largest city in the United States. Park Forest is a village located south of Chicago, in Cook County, Illinois. It has a population of 23,462 human residents (United States Census Bureau 2010). Bourbonnais is a village in Kankakee County, Illinois and has a population of 15,256 human residents (United States Census Bureau 2010). As of April 2010, there is a lack of adequate bedrock geologic knowledge for Kankakee County, Illinois (Leib, Susan E. 2010).

Ecosystems in Illinois include forests, farmland, prairies, wetlands, lakes, urban areas, and rivers. Illinois has a temperate climate with cold snowy winters, and hot wet summers. Average winter temperatures are approximately -5.5°C in the north and 2.7 °C in the south. Mean summer temperatures are 21.1°C in the north and 25 °C in the south. Average rainfall is approximately 91.44 cm per year, and annual snowfall is 93.98 cm for northern Illinois and 154.84 cm central and southern Illinois. During the study period average temperatures of the coldest month (March) to the warmest (May) were approximately 6.6°C to 18.8°C respectively.

*Little Calumet River*

The Little Calumet River study area is at the urban end of the urban-rural gradient and has 175.4 km of river and tributaries. The Little Calumet River originates at the east end of Gary, IN and flows through South Chicago, IL and Gary IN. The majority of the river flow drains in Lake Michigan. The Little Calumet River flows through or borders the towns of Blue Island, Illinois.
Burnham, Dixmoor, Phoenix, Riverdale, Harvey, Calumet City, Lansing, Dolton, and South Holland in Illinois; and Hammond, Munster, Griffith, Highland, Gary, Lake Station, Portage, Burns Harbor, Porter, and Chesterton in Indiana. The city of Chicago, IL with its dense population is located within the northeastern part of Illinois and lies along the southwestern tip of Lake Michigan. The city lies between two rivers, -the Chicago and the Calumet Rivers. The city occupies an area of 606.2km² and according to the US Census Bureau has a combined residential and industrial of approximately 9.7 million people. South Chicago from 130th to 138th street east of the Bishop Ford freeway and west of the Little Calumet River is home to almost 1,000 residents. It is .77km² and located northwest of a disposal site, numerous manufacturing plants, former steel mills, waste dumps, and landfills. The Calumet River in the Chicago area is surrounded by houses, factories, and a waste management facility. It is not easily accessible to the surrounding community or visitors as a recreational park facility. Study site 1 (preserve) (Lon 87.83682 W, Lat 41.14431 N) is located east of Interstate 57 and is the Little Calumet boat docking area. The area is surrounded by woodlands and is also directly across from houses and a nearby machinery facility. Study site 2 (residential) (Lon 87.56061 W, Lat 41.64822 N) is located at a Chicago Marina. This site is surrounded by houses, nearby roads, and the river flows directly around a waste management facility.

Thorn Creek

The Thorn Creek study area is located in Park Forest, Illinois approximately 40.2 km south of the city of Chicago and represents the middle location along the urban-rural gradient.
Thorn Creek is a sub-watershed of the larger Chicago watershed. Its tributaries include Butterfield Creek, Deer Creek, and North Creek. 105 km of stream wind through 26936 ha of northeastern Illinois and 777 ha of northwestern Indiana. Most streams of the Thorn Creek watershed flow from southwest to northeast from their origins in Eastern Will County to their confluence with Thorn Creek and continue to the Little Calumet River in southern Cook County. Park Forest can be accessed from Interstate 57. It is surrounded by U.S. Highway 30 to the north, Western Avenue on the east, the Canadian National Railway to the west and Thorn Creek to the south. Thorn Creek is approximately 336 hectares and runs from a preserve where the nature center is located through a residential neighborhood along Monee Road. Thorn Creek Preserve became an Illinois State Nature Preserve in 1978, and consists of floodplain to mesic deciduous forest, wetlands, and tall grass prairie communities. Study site 1 (preserve) of Thorn Creek is located approximately 1.12 km east from the start of the trail near the nature center (Lon 87.69867 W, Lat 41.46032 N). Study site 2 (residential) is located approximately .06 km east of W Steger Road (Lon 87.67148 W, Lat 41.46979 N). Houses are located within .02 km to the east and south of the study site.

Kankakee River

The Kankakee River study area is at the rural end of the gradient. It is located in Bourbonnais, Illinois and is approximately 88.5 km south of the city of Chicago. The Kankakee River, a tributary of the Illinois River is approximately 144.8 km long in northwestern Indiana and northeastern Illinois. The river rises in northwest Indiana, approximately 8.05 km southwest
of South Bend. The river curves westward as it enters Kankakee County in northeastern Illinois. Approximately 4.8 km southeast of the city of Kankakee, the Kankakee River receives the Iroquois River from the south and turns sharply to the northwest for its lower 56.3 km. It joins the Des Plaines River from the south to form the Illinois River, approximately 80.5 km southwest of Chicago. The river can be accessed from Interstate 57 or 55 and Illinois Routes 102 on the north and 113 on the south. Study site 1 (preserve) of the Kankakee River is located at Kankakee River State Park (Lon 87.69867 W, Lat 41.46032 N) with floodplain forest cover. Study site 2 (residential) is located off of US-45, N Kennedy Drive (Lon 87.87680 W, Lat 41.12272 N) with roads 0.5 km to the north, east, and south of the site. Land uses adjacent to the site include a baseball field west of the site, and a construction site (Johnson Downs Construction Inc.). In addition, local communities and hospitals are located within 3.2 km northwest of the site. The Kankakee River is a woodland area and has a vast animal population of fox, coyote, deer, badgers, beavers, turtles, wild turkeys, red-winged black birds, herons, bluebirds, frogs, snakes, dove, woodcock, rabbit, squirrel, raccoon, and many others.

2.2. Experimental design

In this study, an urban-rural gradient based on population density in southern Chicago and the surrounding suburban and rural areas was selected. The urban-rural gradient contained 3 cities in Illinois - Chicago, Park Forest, and Bourbonnais with population densities >386 people/km², 193 people/km², and <193 people/km² for the three cities, respectively. In each study site, two plots which represented residential and preserve areas were chosen. Data was
collected at Little Calumet River (Chicago), Thorn Creek (Park Forest) and Kankakee River (Bourbonnais). Handheld meters were randomly placed and fully submerged in the water at three random locations for each body of water plot biweekly from March to May 2010.

2.3. Water sampling

Samples were collected three times in March 2010 and two times in April 2010 and May 2010 between 8:00am to 1:00 pm CDT. At each of the three locations- Kankakee River, Thorn Creek, and Little Calumet River, samples were collected at preserve and residential sites. For each sample a YSI Model 85 handheld portable meter (YSI Inc, Yellow Springs, OH, USA) was used to measure dissolved oxygen, conductivity, salinity, and temperature of water. A Hanna HI 9812-5 portable multi-meter (Hanna Instruments, Inc., Woonsocket, RI, USA) was used to measure water pH. During each sample collection date at each sampling site, measurements of environmental variables were made at three random locations and averaged. The probe end of each meter was dipped into the sub surface area (approximately 0.001 km) of the water while the digital value was read and recorded. Dissolved oxygen was measured in percent saturation (% sat.), conductivity in micro Siemens per centimeter (µS/cm), salinity in parts per thousand (ppt), and temperature in degrees Centigrade (°C).

2.4. Statistical Analysis

Two analyses were conducted to determine if the environmental parameters were affected by heavily populated residential areas and if water bodies near the urban end of the gradient
differed significantly from the rural end of the gradient. A two-tailed t-test using mean values was performed to test the null hypothesis that there are no differences in dependent variables—dissolved oxygen, temperature, pH, salinity, and conductivity when comparing preserve vs. residential aquatic systems at three locations along an urban-rural gradient. A repeated measures MANOVA analysis conducted on differences between preserve and residential mean values was used to test three null hypotheses that there are no overall differences in environmental variables of aquatic systems at three locations along an urban-rural gradient based on location, date, and (date x location) interaction.

3. Results

Preserve sites had significantly higher salinity values than residential sites (p=0.029) (Fig 2.). There were no significant differences in residential and preserve sites for environmental parameters water pH, dissolved oxygen, temperature, and conductivity. The results showed that average values for parameters in residential areas for temperature ranged 3.45-17.1°C, dissolved oxygen 73.73-112.03 % saturation, conductivity 352.33-1113.33 µS, salinity 0.2-0.5 ppt, and pH 7.8-8.9. Average values for parameters in preserve areas for temperature ranged 3.6-17.23°C, dissolved oxygen 42.67-104.23 % saturation, conductivity 352.95-1299.57 µS, salinity 0.1-0.8 ppt, and pH 7.2-8.87 (Table 1.)
There was a significant difference in all variables along the urban-rural gradient based on location (p<0.001), date (p=0.001), and location date interaction (p=0.001) (Table 2). The values for examined parameters in the study ranged 3.45-17.10°C (temperature), 73.73-112.03 % saturation (dissolved oxygen), 352.33-1113.33 μS (conductivity), 0.20-0.50ppt (salinity), and 7.80-8.90 (pH) in residential areas. While values ranged 3.60-17.23°C (temperature), 42.67-104.23 % saturation (dissolved oxygen), 352.95-1299.57 μS (conductivity), 0.10-0.80ppt (salinity), and 7.20-8.87 (pH) in preserve areas respectively. On average, along the urban-rural gradient there was a decrease in conductivity and salinity in residential areas. Average values for parameters in the Little Calumet River residential site were 10.34±5.55 °C (temperature), 8.18±0.28 (pH), 900.67±162.38 μS (conductivity), 0.49±0.03 ppt (salinity), 86.46±9.29 % saturation (dissolved oxygen)(Table 3). Average values for parameters in Thorn Creek residential site ranged 8.93±4.10°C (temperature), 8.49±0.32 (pH), 522.11±154.60μS (conductivity), 0.34±0.05ppt (salinity), 94.50±12.31 % saturation (dissolved oxygen)(Table 4). Average values for parameters in the Kankakee River residential site ranged 11.23±3.49 °C (temperature), 8.27±0.32 (pH), 531.03±102.37 μS (conductivity), 0.29±0.04ppt (salinity), 86.99±7.36% saturation (dissolved oxygen)(Table 5). On average, along the urban-rural gradient there was a increase in dissolved oxygen, and a decrease in conductivity and salinity. Average values for parameters in the Little Calumet River preserve sites ranged 11.36±3.64 °C (temperature), 8.08±0.51 (pH), 952.57±186.14 μS (conductivity), 0.59±0.13 (salinity), 71.70±21.21 % saturation (dissolved oxygen)(Table 3). Average values for parameters in Thorn Creek preserve sites ranged 8.93±3.99°C (temperature), 8.37±0.26 (pH), 612.68±212.83 μS (conductivity), 0.39±0.09ppt (salinity), 86.91±4.45 % saturation (dissolved oxygen)(Table 4). Average values for parameters in the Kankakee Riverpreserve site ranged 11.27±3.08 °C
(temperature), 8.49±0.24 (pH), 473.90±96.10µS (conductivity), 0.27±0.07ppt (salinity), 87.10±4.74% saturation (dissolved oxygen) (Table 5). The range of environmental parameters for residential and preserve sites in the Little Calumet River, Thorn Creek, and Kankakee River are shown in Tables 6 and 7.

Environmental variables in all three locations were significantly different due to location and date. Residential mean values were subtracted from preserve values to compare differences in preserve and residential sites. This eliminates the residential and preserves plots, thus allowing for comparison of only the urban-rural gradient. Temperature values at the Little Calumet River were higher for preserve sites in early spring and higher for residential sites in late spring. Thorn Creek Values were higher for preserve sites in early spring. (Fig 3.) Dissolved Oxygen values at the Little Calumet River were higher for preserve sites in early and mid spring. Values for residential sites were higher in late spring. Thorn Creek Values were higher at residential sites for early and mid spring. Kankakee values were higher for residential sites in mid spring. (Fig 4.) Conductivity values at the Little Calumet River were higher in preserve sites in early spring and mid spring; residential values were higher in late spring. Values at Thorn Creek were higher at preserve sites in mid spring; values at the Kankakee River were higher for residential sites in mid spring. (Fig 5.) Salinity values at the Little Calumet River site were higher for preserve sites in early and mid spring. Thorn Creek values were higher for preserve sites in early spring. Kankakee values were higher for residential sites in mid spring. (Fig 6.) pH values for the Little calumet River were higher for preserve sites in early spring and higher for residential sites in late spring. Thorn Creek values were higher for residential in early, mid, and late spring. Kankakee values were higher for preserve sites in early and mid spring (Fig 7.).
Fig 2. Average salinity values for preserve and residential sites

Table 1. Range of environmental parameters values for residential and preserve plots*

<table>
<thead>
<tr>
<th>Plots</th>
<th>T (°C)</th>
<th>DO (% sat)</th>
<th>Con (µS)</th>
<th>Sal (ppt)</th>
<th>pH</th>
</tr>
</thead>
<tbody>
<tr>
<td>Residential</td>
<td>3.45-17.1</td>
<td>73.73-112.03</td>
<td>352.33-1113.33</td>
<td>0.2-0.5</td>
<td>7.8-8.9</td>
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<tr>
<td>Preserve</td>
<td>3.6-17.23</td>
<td>42.67</td>
<td>352.95-1299.67</td>
<td>0.1-0.8</td>
<td>7.2-8.887</td>
</tr>
</tbody>
</table>

T: Temperature; DO: Dissolved Oxygen; Con: Conductivity; Sal: Salinity;

Table 2. P values-MANOVA analysis
<table>
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<tr>
<th>Source</th>
<th>Wilk’s Lambda (Pr&gt;F)</th>
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</thead>
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<td>Location (Within Subjects)</td>
<td>0.001</td>
</tr>
<tr>
<td>Date (Between Subjects)</td>
<td>&lt;0.001</td>
</tr>
<tr>
<td>Location-Date (Between Subjects)</td>
<td>&lt;0.001</td>
</tr>
</tbody>
</table>

Table 3. Mean environmental parameter values for residential and preserve plots of The Little Calumet River*

<table>
<thead>
<tr>
<th>Plots</th>
<th>T (°C)</th>
<th>DO (% sat)</th>
<th>Con (µS)</th>
<th>Sal (ppt)</th>
<th>pH</th>
</tr>
</thead>
<tbody>
<tr>
<td>Residential</td>
<td>10.34±5.55</td>
<td>86.46±9.29</td>
<td>900.67±162.38</td>
<td>0.49±0.03</td>
<td>8.18±0.28</td>
</tr>
<tr>
<td>Preserve</td>
<td>11.36±3.64</td>
<td>71.90±21.21</td>
<td>952.57±186.14</td>
<td>0.59±0.13</td>
<td>8.08±0.51</td>
</tr>
</tbody>
</table>

T: Temperature; DO: Dissolved Oxygen; Con: Conductivity; Sal: Salinity;

Table 4. Mean environmental parameter values for residential and preserve plots for Thorn Creek*

<table>
<thead>
<tr>
<th>Plots</th>
<th>T (°C)</th>
<th>DO (% sat)</th>
<th>Con (µS)</th>
<th>Sal (ppt)</th>
<th>pH</th>
</tr>
</thead>
<tbody>
<tr>
<td>Residential</td>
<td>8.93±4.10</td>
<td>94.50±12.31</td>
<td>522.11±154.60</td>
<td>0.34±0.05</td>
<td>8.49±0.32</td>
</tr>
<tr>
<td>Preserve</td>
<td>8.93±3.99</td>
<td>86.91±4.45</td>
<td>612.68±212.83</td>
<td>0.39±0.09</td>
<td>8.37±0.26</td>
</tr>
</tbody>
</table>

T: Temperature; DO: Dissolved Oxygen; Con: Conductivity; Sal: Salinity;
Table 5. Mean environmental parameter values for residential and preserve plots for The Kankakee River*

Kankakee River

<table>
<thead>
<tr>
<th>Plots</th>
<th>T (°C)</th>
<th>DO (% sat)</th>
<th>Con (µS)</th>
<th>Sal (ppt)</th>
<th>pH</th>
</tr>
</thead>
<tbody>
<tr>
<td>Residential</td>
<td>11.23±3.49</td>
<td>86.99±7.36</td>
<td>531.03±102.37</td>
<td>0.29±0.04</td>
<td>8.27±0.32</td>
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<tr>
<td>Preserve</td>
<td>11.27±3.08</td>
<td>87.10±4.74</td>
<td>473.90±96.10</td>
<td>0.27±0.07</td>
<td>8.49±0.24</td>
</tr>
</tbody>
</table>

T: Temperature; DO: Dissolved Oxygen; Con: Conductivity; Sal: Salinity;

Table 6. Range of environmental parameter values for Residential sites across the urban-rural gradient*

<table>
<thead>
<tr>
<th>Site</th>
<th>T (°C)</th>
<th>DO (% sat)</th>
<th>Con (µS)</th>
<th>Sal (ppt)</th>
<th>pH</th>
</tr>
</thead>
<tbody>
<tr>
<td>Little Calumet River</td>
<td>3.45-17.1</td>
<td>74.13-96.6</td>
<td>671-1113.3</td>
<td>0.4-0.5</td>
<td>7.8-8.7</td>
</tr>
<tr>
<td>Thorn Creek</td>
<td>4.26-15.16</td>
<td>81.23-112.03</td>
<td>352.3-733.3</td>
<td>0.3-0.4</td>
<td>8.13-8.9</td>
</tr>
<tr>
<td>Kankakee River</td>
<td>7.65-15.86</td>
<td>73.3-93.3</td>
<td>446.4-669.6</td>
<td>0.2-0.3</td>
<td>7.95-8.41</td>
</tr>
</tbody>
</table>

T: Temperature; DO: Dissolved Oxygen; Con: Conductivity; Sal: Salinity;

Table 7. Range of environmental parameter values for Preserve sites across the urban-rural gradient*

<table>
<thead>
<tr>
<th>Plots</th>
<th>T (°C)</th>
<th>DO (% sat)</th>
<th>Con (µS)</th>
<th>Sal (ppt)</th>
<th>pH</th>
</tr>
</thead>
<tbody>
<tr>
<td>Residential</td>
<td>11.23±3.49</td>
<td>86.99±7.36</td>
<td>531.03±102.37</td>
<td>0.29±0.04</td>
<td>8.27±0.32</td>
</tr>
<tr>
<td>Preserve</td>
<td>11.27±3.08</td>
<td>87.10±4.74</td>
<td>473.90±96.10</td>
<td>0.27±0.07</td>
<td>8.49±0.24</td>
</tr>
</tbody>
</table>

T: Temperature; DO: Dissolved Oxygen; Con: Conductivity; Sal: Salinity;
<table>
<thead>
<tr>
<th>Site</th>
<th>T (°C)</th>
<th>DO (% sat)</th>
<th>Con (µS)</th>
<th>Sal (ppt)</th>
<th>pH</th>
</tr>
</thead>
<tbody>
<tr>
<td>Little Calumet</td>
<td>6.6-17.2</td>
<td>42.6-104.2</td>
<td>739.6-1299.6</td>
<td>0.4-0.8</td>
<td>7.2-8.6</td>
</tr>
<tr>
<td>River</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Thorn Creek</td>
<td>3.6-15.0</td>
<td>81.3-91.2</td>
<td>352.9-917.6</td>
<td>0.3-0.5</td>
<td>8-8.6</td>
</tr>
<tr>
<td>Kankakee River</td>
<td>8.13-15.3</td>
<td>81.2-94.4</td>
<td>400.4-629.3</td>
<td>0.1-0.3</td>
<td>8-8.8</td>
</tr>
</tbody>
</table>

T: Temperature; DO: Dissolved Oxygen; Con: Conductivity; Sal: Salinity;

---

**Temperature Gradient Comparison**

Fig3. Temperature variance along the urban-rural gradient over time (±S.E.)
Fig 4. Dissolved Oxygen variance along the urban-rural gradient over time (±S.E.)

Fig 5. Conductivity variance along the urban-rural gradient over time (±S.E.)
Fig 6. Salinity variance along the urban-rural gradient over time (±S.E.)

Fig 7. pH variance along the urban-rural gradient over time (±S.E.)
Fig 8. Preserve and residential mean DO values over time along the gradient (means ±S.E.)
Fig 9. Preserve and residential mean Cond values over time along the gradient (means ±S.E.)
Fig 10. Preserve and residential mean Sal values over time along the gradient (means ±S.E.)
Fig 11. Preserve and residential Temp values over time along the gradient.
Fig 12. Preserve and residential pH values over time along the gradient
Table 8 Comparison of results to EPA water quality standards*

<table>
<thead>
<tr>
<th></th>
<th>T(°C)</th>
<th>DO (% sat)</th>
<th>Con (µS)</th>
<th>Sal (ppt)</th>
<th>pH</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Lowest</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>March</td>
<td>3.5 (Cal)</td>
<td></td>
<td>41.5 (Cal)</td>
<td>336.4 (Cal)</td>
<td>.1 (KK)</td>
</tr>
<tr>
<td>May</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>7.9 (KK)</td>
</tr>
<tr>
<td><strong>Highest</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>May</td>
<td>17.3 (TC)</td>
<td></td>
<td>112.7 (TC)</td>
<td>1301 (Cal)</td>
<td>.8 (Cal)</td>
</tr>
<tr>
<td>March</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td>9.0 (TC)</td>
</tr>
<tr>
<td><strong>Res (avg)</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>March</td>
<td>10.38±4.26</td>
<td>89.07±10.17</td>
<td>653.80±218.37</td>
<td>0.37±0.09</td>
<td>8.32±0.31</td>
</tr>
<tr>
<td>April</td>
<td>10.64±3.62</td>
<td>81.78±14.31</td>
<td>686.08±259.60</td>
<td>0.41±0.16</td>
<td>8.30±0.40</td>
</tr>
<tr>
<td><strong>EPA Criteria</strong></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td>0-32 °C</td>
<td>80-120%</td>
<td>0-800µS/cm</td>
<td>0-.5ppt</td>
<td>5-9</td>
</tr>
</tbody>
</table>

T: Temperature; DO: Dissolved Oxygen; Con: Conductivity; Sal: Salinity; Res (avg): Residential averages; Pre (avg): Preserve averages; Cal: Little Calumet River; TC: Thorn Creek; KK: Kankakee River

4. Discussion

In all natural ecosystems such as streams and rivers changes are likely to occur through time. Over a geologic scale major changes occur, over a small scale such as a human lifespan changes are so gradual that they are nearly undetectable. When an ecosystem is disturbed due to human activity, these changes are accelerated. One of the most common changes is alterations in the volume of water flowing through a river or stream channel. In this study preserve sites had higher salinity volumes than residential ones. This may be due to construction of impervious surfaces such as nearby parking lots, which alter the direction of stream flow. Residential sights may have also had a downstream flow which ultimately results in an increase in water volume. Although there were no significant differences in residential and preserve sites for the environmental parameters except for salinity, measurements across the urban-rural gradient
indicate strong variations associated with the Little Calumet River’s dissolved oxygen, conductivity, salinity (Fig 8-10).

The Calumet River showed the most instability, values varied significantly from the Kankakee River and Thorn Creek over a period of time. There was a major decrease in dissolved oxygen at the preserve site in May (Fig 8). And during the study period the Little Calumet River had the highest conductivity and salinity values when compared to Thorn Creek and the Kankakee River (Fig 9-10). It is suggested that urbanization has had an effect on this particular aquatic system which have shown to have increased levels of dissolved solids and the lowest photosynthetic activity from plants and animals. The Kankakee River had slight variations but overall proved to be more stable than the other two sites. The Kankakee River had a balance within all parameters and over time didn’t have any extreme measurements. Environmental parameters for this aquatic system proved to be less affected by human activity. The rural aquatic system doesn’t suffer from the dangers of urbanization. Thorn Creek had the highest dissolved oxygen, temperature, and pH measurements (Fig 8., 11.-12.), this may be due to the lack of cover in the canopy surrounding the creek. Sunlight and rain was better able to reach the surface of the water, therefore photosynthetic activity from aquatic plants and animals were heightened causing an increase in dissolved oxygen levels. This particular suburban site had nearby roads and lacked local manufacturing companies. This may account for the high pH levels in the creek due to the lack of atmospheric pollution and the amount of unpolluted precipitation. Kankakee River had the lowest salinity and pH measurements.
Aerial photographs of the three locations (Fig 13.) illustrate changes in land cover along the gradient. The amount of land cover is directly proportional to the amount of urbanization, thus more variation and extreme measurements were found in the Calumet River (lowest temperature, lowest dissolved oxygen, lowest and highest conductivity, and highest salinity—(Table 8.) when compared to the other two locations. The density of roads might be useful for a study of the dispersal of mammals in an urban area, but a map of land cover is more useful for a study of changes in water bodies along a gradient (Hahs and McDonnell 2006).
Many studies on eutrophication (Beeton 1965), sulphur contamination of rivers (Berner 1971), and general degradation of river quality (Wolman 1971), have been conducted on urbanization and how human pressures are leading to global change in aquatic systems. The results from this analysis over a three month period within less than a 96km gradient revealed a great amount of variance within the aquatic systems, which prove that the security of future water resources is threatened and that urbanization does in fact play a role. The hydrological balance is dependent on transfer of river material, such as carbon and nutrient balance, and biodiversity (Meybeck 2003), so one ecosystem can create global effects. Due to the amount of variation at the Little Calumet River, it can be concluded that the water resources are not exploited in an optimal way. Surface water transfers from the Little Calumet River are modified through extensive land cover (urbanization) and past and present industrialization from nearby factories, power plants, and waste facilities (Fig 14). According to Meybeck 2003, one of the most common uses of water is still dilution and downstream transfer of waste. Deep saline ground water is pumped and connected to surface water increasing river fluxes of nutrients and pollutants. Salinization, which was found to have a significant difference in preserve sites and in the Little Calumet River for this study, results from increase of sodium, chloride, and sulfate due to industrial and mining wastewaters or from poor irrigation practices (Chilton 1989). This can cause water use limitations and is indicative of early warning degradation of water quality.
Salinity has been viewed as one of the most important variables influencing the utilization of organisms in aquatic systems. The Little Calumet River had the most and only extreme values which fell either below or above EPA criteria (Table 8). Due to high salinity and conductivity readings at the Little Calumet River, dissolved oxygen concentrations were low. With increased salinity and conductivity, dissolved oxygen in water decreases. The presence of salts limits the amount of oxygen that can dissolve in water, and can be indicative of water pollution and contamination. Freshwater holds more oxygen than salt water, so the oxygen content can affect the reaction rates of aquatic plants and animals. Dissolved oxygen depletion could suppress respiration, cause death of fish, depress feeding behavior, and affect embryonic development. Water temperature correlated significantly with pH, but has no significant correlation with salinity, conductivity, and dissolved oxygen. From March to May as water temperature increased there was a decrease in pH. This may be due to the rainy months and the
amount of acid rain from sulfur and nitrogen atmospheric pollution. The results from the statistical analysis showed significant correlations between the variables at the three different aquatic systems.

5. Conclusion

The analysis emphasize that aquatic systems are the product of their land, and the influence of aquatic systems is driven by human pressures. If an ecosystem cannot sustain plant and lower animal species life, eventually human life will become unsustainable. Urban areas provide a gateway for invasive species, which result in increased erosion, elimination of vegetation, and removal of food sources for many native species. It is no doubt that aquatic systems of this era are no longer controlled by only Earth system processes. Aquatic system variation is directly influenced by flood regulation, fragmentation, sediment balance, neo-arheism, salinization, chemical contamination, acidification, eutrophication, and microbial contamination. As a result, measures of urbanization capture the greatest variability within aquatic systems and are indicative of water quality output.

The findings indicate that efforts should be stressed to preserve and improve land in order to achieve a hydrological balance. Even though there are organizations such as the Environmental Protection Agency (EPA) and Illinois Conservation Reserve Program (ICR) urban aquatic
systems still suffer from contamination and pollution. Legal history investigations have found that penalties for failure to meet environmental regulations are lower in urban areas. Although discharges have been reduced, contaminants continue to impair the area. And delays of site approval in low income areas are longer as well. This research has revealed an imbalance at the Little Calumet River test site. Further research should be proposed to determine how to reduce pollution in and around the River, and enforce water regulation laws. Future research should demonstrate ways to correct the water quality balance in order to make it available for recreational activities and aquatic survival.

There should be a greater focus on preventive management and remediation engineering solutions. These goals should include waste minimization, resource recovery, pollution prevention, and consideration of environmental effects during product development. Future environmental health programs should be strongly considered and emphasize efficient monitoring of environmental quality, interpretation of the impact of human actions on terrestrial and aquatic ecosystems, and a urgent need to comply with environmental laws and regulations regarding ground water decontamination and clean air.

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